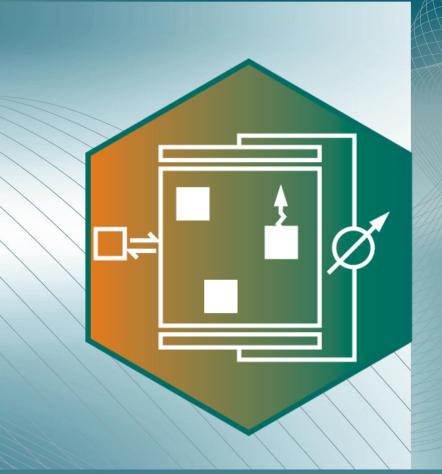
# **Towards Robust Membranes Based On Poly-Electrolytes**

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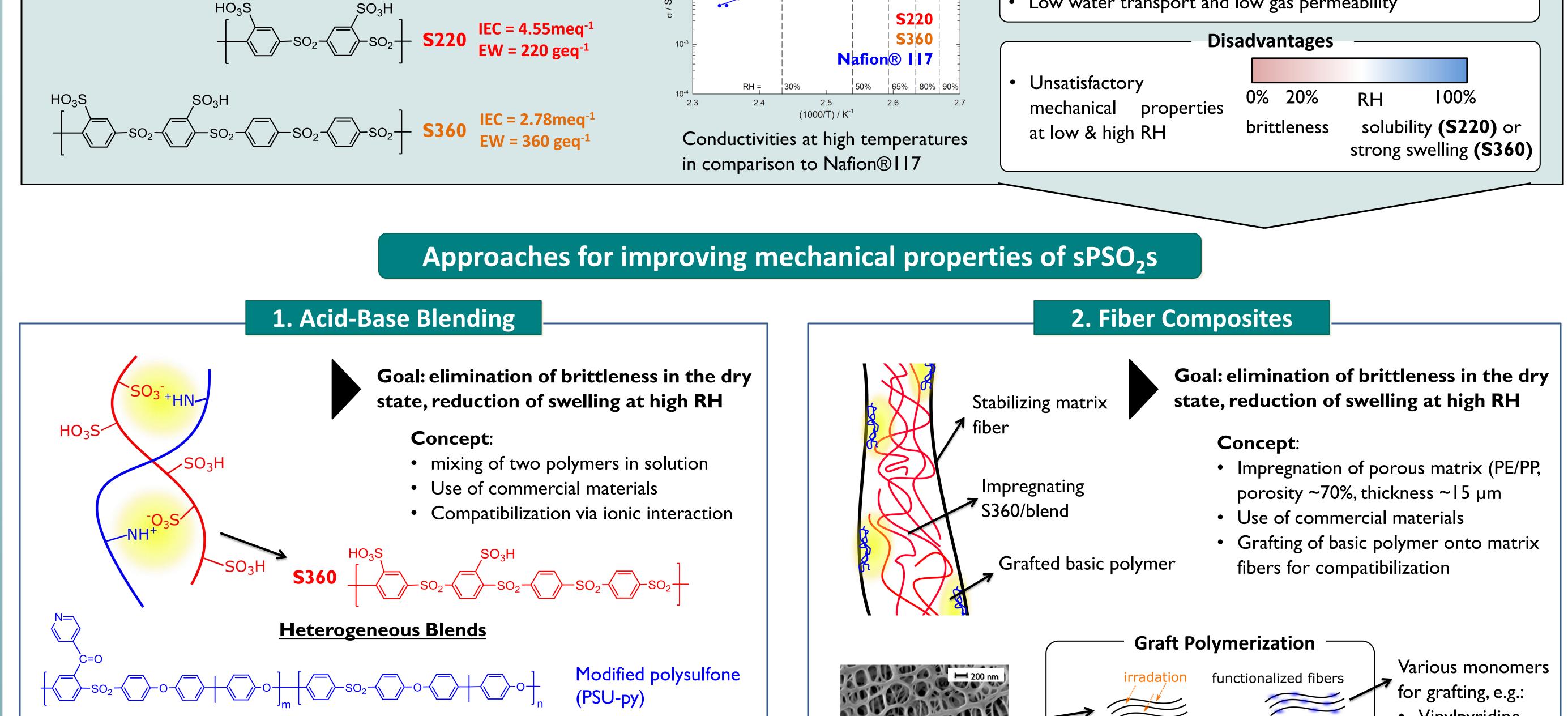
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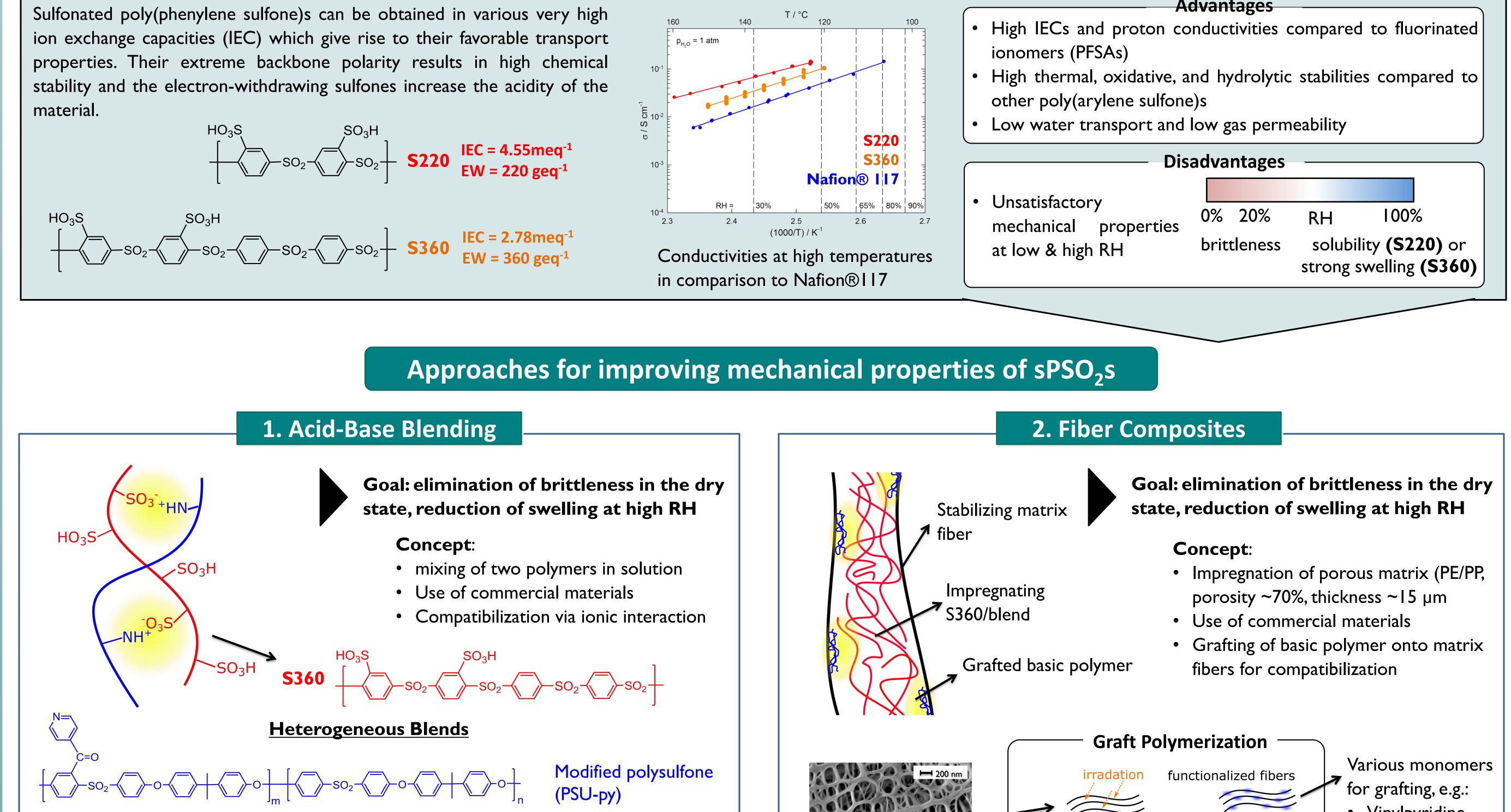


## Introduction

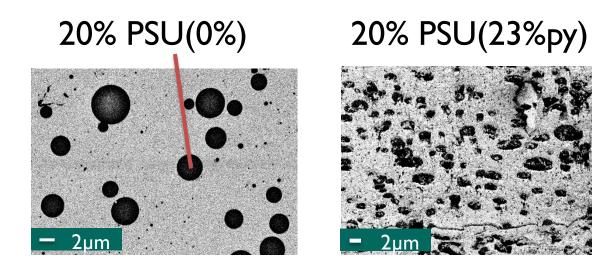
Proton-exchange membranes (PEMs) are used in fuel cells. Commercial PEM materials are primarily perfluorosulfonic acid-based ionomers (PFSAs, e.g. Nafion®). Under fuel cell operating conditions at high temperature (>90°C) and low relative humidity (RH), PFSAs show a significant drop in conductivity, limiting the overall performance. Sulfonated poly(phenylene sulfone)s (sPSO<sub>2</sub>s) with their exceptionally high conductivities could overcome these limitations, if their mechanical weakness at low and high RH can be reduced.

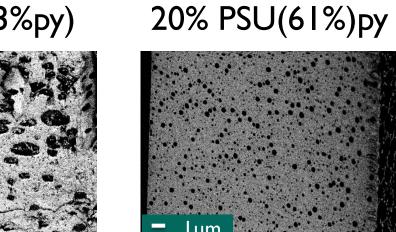
## Sulfonated poly(phenylene sulfone)s<sup>[1][2][3]</sup>

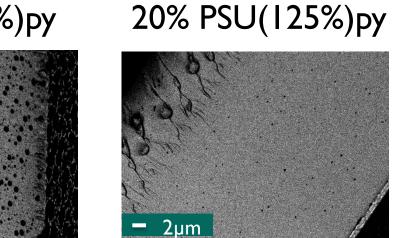




#### **Advantages**

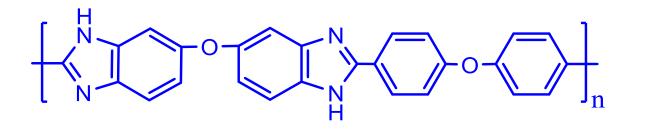






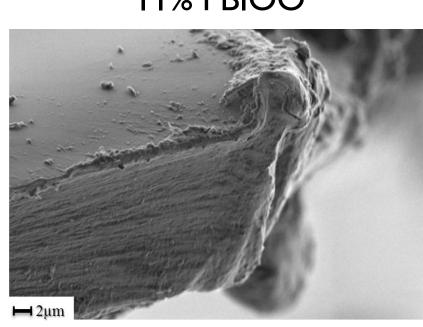
- Loss of conductivity mainly due to volume effect of PSU-py
- High pyridine-modification results in more even distribution of PSU-phase
- High modification changes micro-phase separation to nano-phase separation
- Trade-off between
  - improved blending behavior
  - decreased mechanical stability of PSU (PSU-125%py itself is brittle)

#### **Homogeneous Blends**



Polybenzimidazole (PBIOO) I I% PBIOO

- High loss of conductivity due to ionic cross-linking (reduced IEC)
- Benzimidazole is stronger base than pyridine
- Small EW of PBIOO  $\rightarrow$  low amounts (wt%) possible
- Neutralization of S360 necessary for blending



### Mechanical Analysis - First Results

• Micro-phase separated blends provide only small, insystematic stabilization

• Vinylpyridine polymerization • Vinylimidazole Structure of porous • Control of grafting degree via temp., matrix time and monomer concentration **S360-composite Fuel Cell Test First Results** Performance similar Easy thin membrane T = 80 °C to Nafion® 112 preparation RH = 100% Stabilizing effect at low FC-test conducted on ≥ 0.6temp. (limit of PE) non-grafted Significant swelling composite (matrix + T = 95 °C T = 95 °C RH = 70%**RH = 70%** S360) (at high RH) 0.2 -Impregnation control p = 2.5 bar crucial 2000 500 1000 1500 current density /mAcm<sup>-2</sup>

## References

[1] Schuster, M.; Kreuer, K.D.; Andersen, H.T.; Maier, J., Macromolecules 2007, 40, 598. [2] Schuster, M.; De Araujo, C.C.; Atanasov, V.; Andersen, H.T.; Kreuer, K.D.; Maier, J., Macromolecules 2009, 42, 3129. [3] Wohlfarth, A., Dissertation, Universität Stuttgart, 2015.

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